

Novel Chemoenzymatic Synthesis of Peptide C-Terminal Amides from Ester Precursors

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Abstract: A novel two step chemoenzymatic procedure for the preparation of peptide C-terminal amides has been developed.

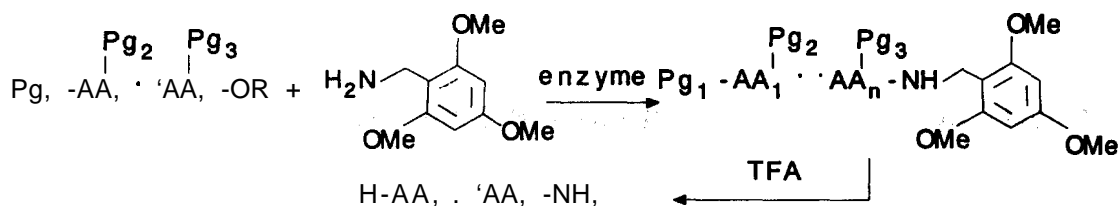
Recent advances in multiple peptide synthesis make possible the simultaneous preparation and analysis of very large peptide libraries. It is due to this progress, that in the nearest future we may see a dramatic increase in the number of peptide-based pharmaceuticals, such as neuropeptide hormones.² The fact that almost half of all known neuropeptides are C-terminal amides demonstrates the importance of the carboxamido group for full biological activity of these compounds.³

The preparation of substantial amounts of these peptides can be better achieved by solution-, solid-phase peptide synthesis, or enzymatic techniques. Each of these methods has its own advantages and limitations. In solution-phase peptide synthesis, the preparation of peptide C-terminal amides often suffers due to the poor solubility of amide intermediates in commonly used organic solvents. This can be overcome by use of C-terminal esters and ammonolysis with ammonia at the conclusion of the synthesis. However, this reaction is often slow, low yielding, prone to racemization and lacks selectivity when other esters (eg. benzyl) are present in the molecule.⁴ Solid phase synthesis allows the preparation of peptide amides but the high cost of the required resins restricts the adaptability of this technique especially on a large scale. Amidation can also be achieved by enzyme-catalyzed hydroxylation of C-terminal Gly residues, transpeptidation or aminolysis.⁶ These reactions normally employ peptide precursors, the sequences of which are either one amino acid shorter⁷ or contain an alternative C-terminal amino acid to the desired final product.⁸

Here we report a novel two step chemoenzymatic method for the preparation of C-terminal amides from the peptides of desired length.⁹ First, regioselective enzyme-mediated coupling of the corresponding peptide ester to DmbNH₂ or TmbNH₂¹⁰ is carried out in aqueous-organic solvent mixtures or in organic solvents using papain or subtilisin. Second, the resulting product is treated with trifluoroacetic acid to give a fully deprotected peptide C-terminal amide (see Figure).

The resulting peptide di- and tri-methoxymethylamides were isolated in good yield and high purity after a simple work-up procedure.¹¹ In addition, papain-catalyzed reactions were fast and complete in less than 2 h. It is worth mentioning that different esters, such as methyl, 2-chloroethyl and plienacyl were successfully employed.

To demonstrate the versatility of this technique, ZGlu(^tBu)Lys(Boc)NHTmb, a peptide which, in addition to the benzylamide moiety, contains three of the most widely used protecting groups was prepared on a large scale (1.1 g). This compound was treated with TFA/thioanisole to give the desired fully deprotected C-terminal amide (HGluLysNH₂) in quantitative yield.¹²



Peptide ester	Enzyme	Time	Yield (%)
ZLeuLys(Z)OMe	Papain	20 min	76
BocLeuGlu(Bn)OPac	Papain	65 min	69
BocMetLeuOMe	Papain	60 min	63
ZGlu(^t Bu)Lys(Boc)OMe	Papain	2h	87
ZPheOCH ₂ CH ₂ Cl ^a	Subtilisin	24h	64

a. Reaction was carried out in *tert*-amyl alcohol with DmbNH₂ (3 eq.)

References and Notes

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- In a typical experiment protected peptide ester (80 mg) and TmbNH₂ hydrochloride (3eq.) are suspended in 6-8 mL 0.2 M carbonate buffer pH 9 containing 40% MeCN. 2-Mercaptoethanol (25 μL) is added followed by papain (5-10 mg) and the reaction stirred at RT (20-120 min). Water is added to complete precipitation of the product, which is filtered, washed with water, and dried.
- Nonstandard abbreviations: Dmb, 2,4-dimethoxybenzyl; Tmb, 2,4,6-trimethoxybenzyl; TFA, trifluoroacetic acid; Pg, protecting group; AA, amino acid,
- All compounds were characterized by TLC, HPLC, ¹H NMR and MS.
- ZGlu(^tBu)Lys(Boc)NHTmb (1 mmol) was stirred in TFA (21 mL), thioanisole (5.9 mL) and H₂O (2.1 mL) overnight at RT. Solvent was evaporated and residue taken up in H₂O, washed with Et₂O and lyophilized. In addition to standard analyses (Ref. 10) HGluLysNH₂ was characterized by FAB MS and high resolution MS.

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